# **MATERIALS 2**

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# Solid, Liquid or Gas?

Solids, liquids and gases are called the three states of matter

Look at the materials you have been given and decide what state they are (solid, liquid or gas) they are. Write the name of each substance in the correct column

Substances you may be given include: ice, water, air, sand, rubber, chocolate, sugar, fizzy drink (the liquid part), fizzy drink (the bubbles of carbon dioxide)

Solid	Liquid	Gas

#### Facts about solids liquids and gases

Solids	Liquids	Gases
Solids do not flow. They always have a fixed shape	Liquids flow and can be poured easily	Gases flow and change shape
Solids always take up the same amount of space	Liquids change their shape depending on their container	Gases spread out to fill the space of their container
Solids can be cut or shaped Solids stay in one place and can be held	Liquids always take up the same amount of space (the <b>volume</b> does not change)	Gases can be squashed (compressed)

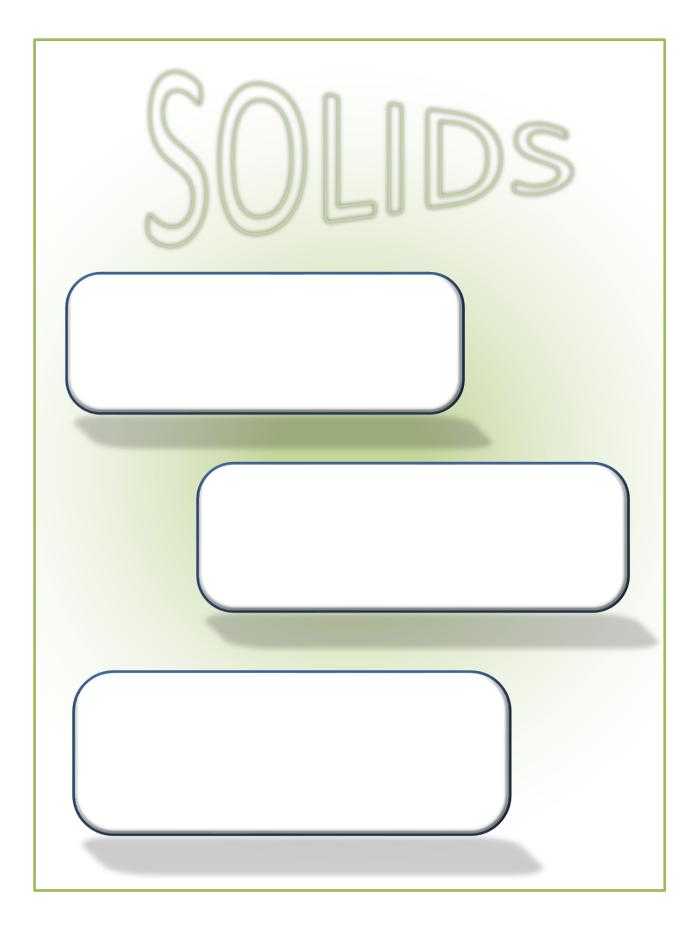
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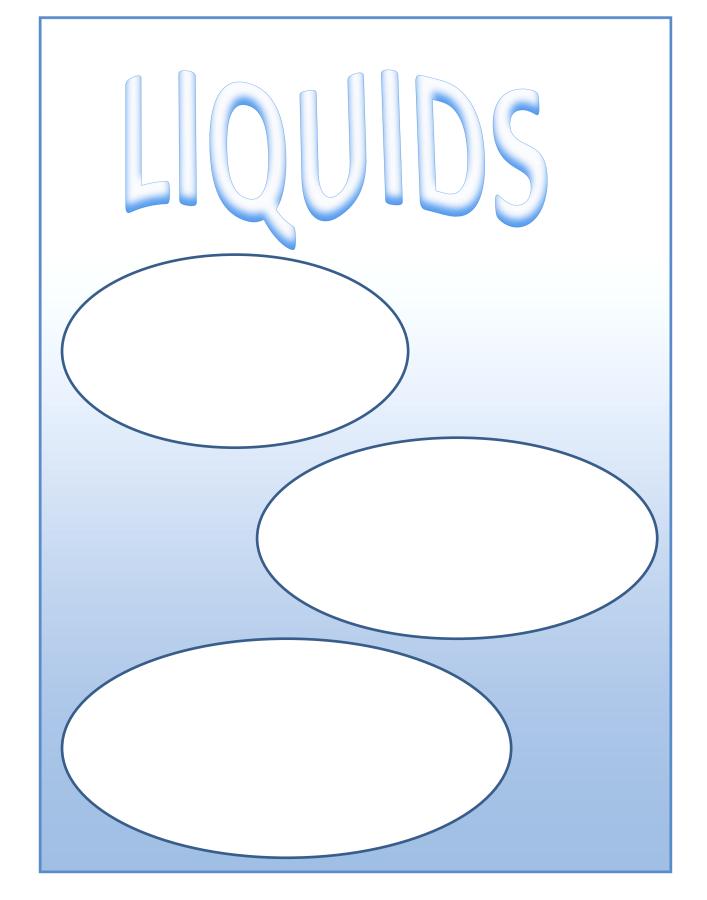
# The three states of matter Summary

	SOLIDS	LIQUIDS	GASES
Examples		Oil	Air
Does it keep its shape?		No	
Does it always have the same volume?		Yes	
Can it flow down a pipe?	No		Yes
Can it be squashed up (can it be compressed?)	No	No	



I can flow easily but always have the same volume	What am I?
I always completely fill whatever container I am put into	What am I?
I am hard and always have the same volume.	What am I?
I can be squashed into smaller spaces	What am I?
I keep my shape and cannot flow	What am I?
I take on the shape of my container but I cannot be squa	shed into a smaller space
	What am I?







## Experiment:

Make your own notes on a separate sheet of paper

## **Making a Solution**

- 1. Place 100 ml of water into a 250ml beaker and add 1 spatula of sand. Stir the mixture well. What happens?
- 2. Place 100 ml of water into a 250ml beaker and add 1 spatula of sugar Stir the mixture well. What happens?

Write a note explaining what you did along with the results. Explain what you observed.

Whenever something dissolves a **solution** is formed

The liquid we used to make the solution is called the **solvent** 

The solid that we dissolved is called the **solute** 

#### Summary:

SUGAR + WATER = SUGAR SOLUTION solute + solvent = solution

Note: The solvent is usually water but could be any liquid like petrol

# Which substances dissolve?

You are now going to do an experiment to see which substances are soluble in water.

1. Place 50 ml of water into a 100ml beaker and add 1 spatula of powdered solid. Stir the mixture well and write down what you observe

#### **RESULTS**

Substance	Observation	Soluble / Insoluble
Salt		
Chalk		
Sand		
Sugar		
Coffee		

#### To make it a fair test I kept these factors the same for each experiment

- The quantity of solid being used.
- The volume of water
- The rate of stirring
- The temperature of the water.
- The size of the particles in the solid

Did all the materials dissolve equally easily?
Out of the materials that DID dissolve, which dissolved the best and which dissolved the least easily?

# Experiment:

Make your own notes on a separate sheet of paper

# Dissolving a sugar lump

Aim:	Investigating	how to	make a	sugar	lump	dissolve	quicker
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#### What you do:

Dissolve a sugar lump using a variety of techniques and measure how long it takes the sugar to dissolve in each case

Remember use the same mass of sugar and same volume of water for each experiment and to record your results carefully

#### **Apparatus needed**

1 x 250ml beaker,	1 spatula,	1 stopwatch or timer	, 1 mortar and	pestle

#### **RESULTS**

Conditions	Time taken to dissolve
A No stirring+ cold water	
B Stirring the lump+ cold water	
C Stirring + crushing + cold water	
D Stirring + crushing + warm water	
E Stirring + warm water (not crushed)	

The experiment where the sugar dissolved the quickest is
I know this because the time taken for the sugar to dissolve was
For each experiment this is what I measured:
This is what I kept the same:
Conclusion: we found three ways to help the sugar lump dissolve quicker:
1
2
3

Assessment test 1 A	Date:
	Datominimi

#### 1. Solid, liquid or gas?

I can flow easily but cannot be squashed up. What am I?. Solid/Liquid/Gas
I always keep my shape and don't change size. What am I? Solid/Liquid/Gas
I always completely fill whatever container I am put in. What am I? Solid/Liquid/Gas

#### **2.** Fill in the table below.

Put the following substances into its correct box below: Air, Water, Brick, Helium, ice, Ink

There should now be two words in each box

Solid	Liquid	Gas
1.	1.	1.
2.	2.	2.

### Making a solution

Sugar easily dissolves in water to form sugar solution Sand does not dissolve

- 1. If something can <u>dissolve</u> we say it is **soluble insoluble**
- 2. If something <u>cannot dissolve</u> then we say it is **soluble insoluble**
- 3. When something dissolves it forms a solution solute solvent
- 4. Name a substance that is **soluble** in water **sand sugar flour**
- 5. Name a substance that is *insoluble* in water sand salt sugar
- 6. Sugar takes a long time to dissolve in cold water

To make the sugar dissolve quicker this is what I would do:						

Assessment test 1B	Date:
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1. Solid, liquid or gas?

I can flow easily but cannot be compressed. What am I?. Solid/Liquid/Gas

I always keep my shape and have a definite volume. What am I? Solid/Liquid/Gas

I can be compressed and always fill whatever container I am put in. What am I?

Solid/Liquid/Gas

**2.** Fill in the table below.

Put the following substances into their correct columns Air, Water, Helium, Brick, Ice, Ink

Write one more substance of your own in each column (there should now be two substances in each)

Solid	Liquid	Gas

## **Making a solution**

Sugar easily dissolves in water to form sugar solution Sand does not dissolve

- 7. If something can dissolve we say it is soluble insoluble
- 8. If something cannot dissolve then we say it is **soluble insoluble**
- 9. When something dissolves it forms a **solution solute solvent**
- 10. Name the **solute** in sugary water **salt sugar water**
- 11. Name the **solvent** in sugary water **sand sugar water**
- 12. Name a substance that is **soluble** in water **sand sugar flour**
- 13. Name a substance that is *insoluble* in water sand salt sugar

	Describe three things the girl could do to help the sugar dissolve more quickly
i.	
ii.	
iii.	

A boy did an experiment to investigate how quickly a sugar lump dissolved.

He took three sugar lumps and dissolved each a different way. He timed how long it took each to dissolve and wrote down his results

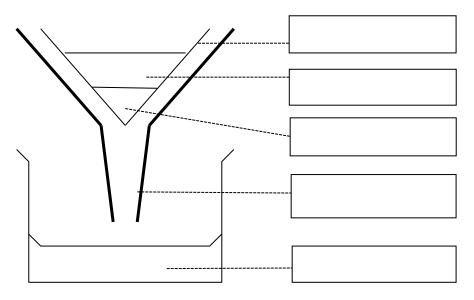
	How he dissolved the lump	Time taken for the sugar lump to dissolve
Sugar Lump 1	Stirred it into cold water	25 seconds
Sugar Lump 2	Stirred it into hot water	12 seconds
Sugar Lump 3	Crushed the lump and then stirred it into cold water	15 seconds

9.	Which sugar lump dissolved the quickest?
10	Which sugar lump dissolved the slowest?

#### How to filter a liquid

Date:.....

Filtering (or filtration) is used to separate a solid from a liquid

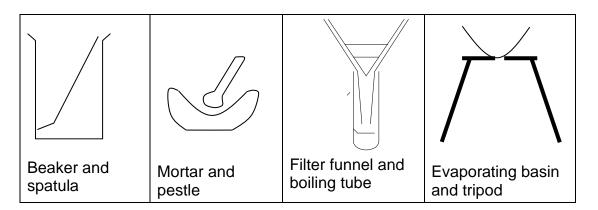


Filter paper, Dirty water, Dirt, Filter funnel, Clear water

Clear water passes through tiny holes in the filter paper and collects in the container below. Particles of dirt do not fit through the holes and remain in the filter paper.

**Note:** The clear water that collects may not be pure. Anything dissolved in the water (like salt) will **not** get stopped by the filter paper neither will very tiny objects like germs.

## **Making Rock salt pure**



Fill in the boxes to show how you made pure salt from rock salt
What we did first:
To do this we used this apparatus:
We then
To do this we used this apparatus
We then
To do this we used this apparatus
We then
To do this we used this apparatus

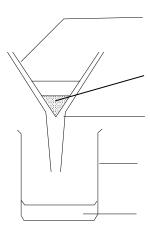
At the end of the experiment we were left behind with:.....

#### Use these sentences to help you fill in the boxes.

- 1. Stirred the rock salt into warm water to dissolve the salt
- 2. Crushed the salt to make it dissolve easier
- 3. Filtered the mixture to remove the dirt
- Warmed the mixture to evaporate the water 4.

A pupil filtered some muddy water using the apparatus below

1. Label the apparatus using words from the list below the diagram

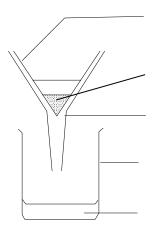


FILTER FUNNEL CLEAR WATER BEAKER DIRT FILTER PAPER

2.	What stays in the filter paper?
3.	What liquid drips into the beaker?
4.	Would the clear water in the beaker be safe to drink? (yes or no)
	Give a reason for your answer above
5.	Why could you not use a sieve to separate mud from water?

A pupil filtered some muddy water using the apparatus below

**6.** Label the apparatus using words from the list below the diagram



FILTER FUNNEL	CLEAR WATER	BEAKER DIRT	FILTER PAPER
I IL I LN I UNINLL	CLLAN WAILN	BLANLN DINI	TILILN FAFLN

_	\ \ \ /   1   _       1	- FU TDATE : 41-:-	10	
/	What would be tr	P FILLRALE IN THIS	eyneriment /	
	VVII at VVO ala DC ti		CADCIIIICIIL: .	

- 8. What would be the RESIDUE in this experiment? .....

10. Why	could	the pup	il not	use	a si	eve t	o sep	arate	mud	from	water	?

11. How could you obtain the substance in the first column from the mixture in the second column (the answers are below the question)

Substance that you want	Mixture that you have	Method of separation
Gravel	sand and gravel	
iron filings	iron filings and sand	
salt	salty water	
dirt	dirt and water	

Use a magnet, Evaporate the water, Filter the mixture, Use a sieve

#### **Words list**

**Natural** Found in nature. Note made by humans. Wood is natural

**Synthetic** Man-made. Artificial. All plastics are man-made.

**Solute** The substance dissolved in a solution

**Solvent** The liquid used to make a solution (usually water)

**Solution** A clear liquid containing a dissolved solid

**Soluble** Able to dissolve eg sugar is soluble in water

**Insoluble** Cannot dissolve eg sand is insoluble in water

**Evaporate** When a liquid turns into a gas

**Boil** When a liquid turns into a gas at its boiling point

**Condense** When a gas or vapour turns into a liquid (when it cools down)

**Freeze** When a liquid turns into a solid (when it cools down)

Melt When a solid turns into a liquid (when it warms up)

**Filtering** What you do to separate the solid from a liquid using filter paper

**Filtrate** The clear liquid that drips through the filter paper during filtration

**Residue** The solid left on the filter paper after filtering