## MATERIALS 2

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Name $\qquad$

Form

## Solid, Liquid or Gas?

Solids, liquids and gases are called the three states of matter
Look at the materials you have been given and decide what state they are (solid, liquid or gas) they are. Write the name of each substance in the correct column

Substances you may be given include: ice, water, air, sand, rubber, chocolate, sugar, fizzy drink (the liquid part), fizzy drink (the bubbles of carbon dioxide)

| Solid | Liquid | Gas |
| :--- | :--- | :--- |
|  |  |  |
|  |  |  |

Facts about solids liquids and gases

| Solids | Liquids |  |
| :---: | :---: | :---: |
| Solids do not flow. They <br> always have a fixed <br> shape | Liquids flow and can be <br> poured easily | Gases flow and change <br> shape |
| Solids always take up the <br> same amount of space <br> Solids can be cut or <br> shaped | Liquids change their <br> shape depending on their <br> container | Gases spread out to fill <br> the space of their <br> container |
| Solids stay in one place <br> and can be held <br> the same amount of <br> space (the volume does <br> not change) | Gases can be squashed <br> (compressed) |  |

The three states of matter
Summary

|  | SOLIDS | LIQUIDS | GASES |
| :--- | :---: | :---: | :---: |
| Examples |  |  |  |
| Does it keep its <br> shape? |  | Oil | Air |
| Does it always <br> have the same <br> volume? |  | No |  |
| Can it flow down <br> a pipe? | Yes |  |  |
| Can it be <br> squashed up <br> (can it be <br> compressed?) | No |  |  |



I can flow easily but always have the same volume

What am I? $\qquad$

I always completely fill whatever container I am put into What am I?. $\qquad$

I am hard and always have the same volume.

I can be squashed into smaller spaces

I keep my shape and cannot flow

What am I?

What am I?.

What am I? $\qquad$

I take on the shape of my container but I cannot be squashed into a smaller space

## What am I?




## Experiment.

Make your own notes on a separate sheet of paper

## Making a Solution

1. Place 100 ml of water into a 250 ml beaker and add 1 spatula of sand. Stir the mixture well. What happens?
2. Place 100 ml of water into a 250 ml beaker and add 1 spatula of sugar Stir the mixture well. What happens?

Write a note explaining what you did along with the results. Explain what you observed.

Whenever something dissolves a solution is formed
The liquid we used to make the solution is called the solvent
The solid that we dissolved is called the solute

Summary:

| SUGAR | WATER | $=$ | SUGAR SOLUTION |
| :--- | :--- | :--- | :--- |
| solute | $+\quad$ solvent | $=$ | solution |

Note: The solvent is usually water but could be any liquid like petrol

## Which substances dissolve?

You are now going to do an experiment to see which substances are soluble in water.

1. Place 50 ml of water into a 100 ml beaker and add 1 spatula of powdered solid. Stir the mixture well and write down what you observe

## RESULTS

| Substance | Observation | Soluble / Insoluble |
| :---: | :---: | :---: |
| Salt |  |  |
| Chalk |  |  |
| Sand |  |  |
| Sugar |  |  |
| Coffee |  |  |
|  |  |  |
|  |  |  |

To make it a fair test I kept these factors the same for each experiment

- The quantity of solid being used.
- The volume of water
- The rate of stirring
- The temperature of the water.
- The size of the particles in the solid

Did all the materials dissolve equally easily? $\qquad$
Out of the materials that DID dissolve, which dissolved the best and which dissolved the least easily?
$\qquad$
$\qquad$

## Experiment.

## Make your own notes on a separate sheet of paper

## Dissolving a sugar lump

Aim: Investigating how to make a sugar lump dissolve quicker

## What you do:

Dissolve a sugar lump using a variety of techniques and measure how long it takes the sugar to dissolve in each case

Remember use the same mass of sugar and same volume of water for each experiment and to record your results carefully

## Apparatus needed

$1 \times 250 \mathrm{ml}$ beaker, 1 spatula, 1 stopwatch or timer, 1 mortar and pestle

## RESULTS

| Conditions | Time taken to dissolve |
| :--- | :--- |
| A No stirring+ cold water |  |
| B Stirring the lump+ cold water |  |
| C Stirring + crushing + cold <br> water |  |
| D Stirring + crushing + warm <br> water |  |
| E Stirring + warm water (not <br> crushed) |  |
|  |  |

The experiment where the sugar dissolved the quickest is $\qquad$

I know this because the time taken for the sugar to dissolve was $\qquad$

For each experiment this is what I measured: $\qquad$

This is what I kept the same: $\qquad$
$\qquad$

Conclusion: we found three ways to help the sugar lump dissolve quicker:

1. $\qquad$
2. $\qquad$
3. $\qquad$

## Assessment test 1 A

Date:.............................

1. Solid, liquid or gas?

I can flow easily but cannot be squashed up. What am I?. Solid/ Liquid/Gas
I always keep my shape and don't change size. What am I? Solid/ Liquid/Gas
I always completely fill whatever container I am put in. What am I? Solid/ Liquid/Gas
2. Fill in the table below.

Put the following substances into its correct box below: Air, Water, Brick, Helium, ice, Ink
There should now be two words in each box

| Solid | Liquid | Gas |
| :--- | :--- | :--- |
| 1. | 1. | 1. |
| 2. | 2. | 2. |

## Making a solution

Sugar easily dissolves in water to form sugar solution Sand does not dissolve

1. If something can dissolve we say it is soluble insoluble
2. If something cannot dissolve then we say it is soluble insoluble
3. When something dissolves it forms a solution solute solvent
4. Name a substance that is soluble in water sand sugar flour
5. Name a substance that is insoluble in water sand salt sugar
6. Sugar takes a long time to dissolve in cold water

To make the sugar dissolve quicker this is what I would do:

## Assessment test 1B <br> Date:.............................

## 1. Solid, liquid or gas?

I can flow easily but cannot be compressed. What am I?. Solid/ Liquid/Gas
I always keep my shape and have a definite volume. What am I? Solid/ Liquid/Gas
I can be compressed and always fill whatever container I am put in. What am I?
Solid/ Liquid/Gas
2. Fill in the table below.

Put the following substances into their correct columns

Air, Water, Helium, Brick, Ice, Ink

Write one more substance of your own in each column (there should now be two substances in each)

| Solid | Liquid | Gas |
| :--- | :--- | :--- |
|  |  |  |

## Making a solution

Sugar easily dissolves in water to form sugar solution Sand does not dissolve
7. If something can dissolve we say it is soluble insoluble
8. If something cannot dissolve then we say it is soluble insoluble
9. When something dissolves it forms a solution solute solvent
10. Name the solute in sugary water salt sugar water
11. Name the solvent in sugary water sand sugar water
12. Name a substance that is soluble in water sand sugar flour
13. Name a substance that is insoluble in water sand salt sugar

Describe three things the girl could do to help the sugar dissolve more quickly
i.
ii.
iii. $\qquad$

A boy did an experiment to investigate how quickly a sugar lump dissolved.
He took three sugar lumps and dissolved each a different way.
He timed how long it took each to dissolve and wrote down his results

|  | How he dissolved the lump | Time taken for the <br> sugar lump to dissolve |
| :---: | :---: | :---: |
| Sugar Lump 1 | Stirred it into cold water | 25 seconds |
| Sugar Lump 2 | Stirred it into hot water | 12 seconds |
| Sugar Lump 3 | Crushed the lump and then stirred it into cold water | 15 seconds |

9. Which sugar lump dissolved the quickest? $\qquad$
10. Which sugar lump dissolved the slowest?

Filtering (or filtration) is used to separate a solid from a liquid


Filter paper, Dirty water, Dirt, Filter funnel, Clear water

Filtrate: $\quad$ The filtrate is the clear liquid that drips through the filter paper Residue: The residue is any solid that gets caught in the filter paper

What is the residue in this experiment (clear water or dirt)? $\qquad$
What is the filtrate in this experiment (clear water or dirt)? $\qquad$

Clear water passes through tiny holes in the filter paper and collects in the container below. Particles of dirt do not fit through the holes and remain in the filter paper.

Note: The clear water that collects may not be pure.
Anything dissolved in the water (like salt) will not get stopped by the filter paper neither will very tiny objects like germs.
$\qquad$
Beaker and

spatula $\quad$\begin{tabular}{l}
Mortar and <br>
pestle

$\quad$

Filter funnel and <br>
boiling tube

$\quad$


| Evaporating basin |
| :--- |
| and tripod | <br>

\hline
\end{tabular}

Fill in the boxes to show how you made pure salt from rock salt

## What we did first:

To do this we used this apparatus:

## We then

To do this we used this apparatus

## We then

To do this we used this apparatus

## We then

To do this we used this apparatus

At the end of the experiment we were left behind with:

## Use these sentences to help you fill in the boxes.

1. Stirred the rock salt into warm water to dissolve the salt
2. Crushed the salt to make it dissolve easier
3. Filtered the mixture to remove the dirt
4. Warmed the mixture to evaporate the water

A pupil filtered some muddy water using the apparatus below

1. Label the apparatus using words from the list below the diagram


FILTER FUNNEL CLEAR WATER BEAKER DIRT FILTER PAPER
2. What stays in the filter paper?
$\qquad$
3. What liquid drips into the beaker?
$\qquad$
4. Would the clear water in the beaker be safe to drink? (yes or no).

Give a reason for your answer above $\qquad$
5. Why could you not use a sieve to separate mud from water?

## Science Assessment Test 2B

Date: $\qquad$

A pupil filtered some muddy water using the apparatus below
6. Label the apparatus using words from the list below the diagram


FILTER FUNNEL CLEAR WATER BEAKER DIRT FILTER PAPER
7. What would be the FILTRATE in this experiment? $\qquad$
8. What would be the RESIDUE in this experiment? $\qquad$
9. Would the clear water be safe to drink? (yes or no) $\qquad$
Give a reason for your answer above $\qquad$
10. Why could the pupil not use a sieve to separate mud from water?
$\qquad$
11. How could you obtain the substance in the first column from the mixture in the second column (the answers are below the question)

| Substance that <br> you want | Mixture that you have | Method of separation |
| :--- | :--- | :--- |
| Gravel | sand and gravel |  |
| iron filings | iron filings and sand |  |
| salt | salty water |  |
| dirt | dirt and water |  |

Use a magnet, Evaporate the water, Filter the mixture, Use a sieve

## Words list

| Natural | Found in nature. Note made by humans. Wood is natural |
| :--- | :--- |
| Synthetic | Man-made. Artificial. All plastics are man-made. |
| Solute | The substance dissolved in a solution |
| Solvent | The liquid used to make a solution (usually water) |
| Solution | A clear liquid containing a dissolved solid |
| Soluble | Able to dissolve eg sugar is soluble in water |
| Insoluble $\quad$ Cannot dissolve eg sand is insoluble in water |  |
| Evaporate $\quad$ When a liquid turns into a gas |  |
| Boil | When a liquid turns into a gas at its boiling point |
| Condense | When a gas or vapour turns into a liquid (when it cools down) |
| Freeze | When a liquid turns into a solid (when it cools down) |
| Melt | When a solid turns into a liquid (when it warms up) |
| Filtering | What you do to separate the solid from a liquid using filter paper |
| Filtrate | The clear liquid that drips through the filter paper during filtration |
| Residue | The solid left on the filter paper after filtering |

